organic compounds

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

(5-Benzoyl-2-methyl-4-{[1-(pyridin-4-yl)-1*H*-1,2,3-triazol-4-yl]methoxy}-1-benzofuran-7-yl)(phenyl)methanone

Xiao-qin Zhang,^a Hai-Liang Zhang,^b Zhu-Yong Dong,^b Qiang Qian^b and Yu-Guang Wang^a*

^aCollege of Biological and Environmental Engineering, Zhejiang University of Technology, Hangzhou 310014, People's Republic of China, and ^bZhejiang SiXian Pharmaceutical Co. Ltd, ShaoXing 312065, People's Republic of China Correspondence e-mail: yuguangw@zjut.edu.cn

Received 13 April 2012; accepted 24 May 2012

Key indicators: single-crystal X-ray study; T = 296 K; mean σ (C–C) = 0.006 Å; R factor = 0.053; wR factor = 0.200; data-to-parameter ratio = 13.3.

The crystal structure of the title compound, $C_{31}H_{22}N_4O_4$, features weak $C-H\cdots O$ interactions. The dihedral angle between the fused benzene and furan rings is 2.49 (15)°, while that between the triazole and pyridine rings is 10.23(18)°.

Related literature

For bioactive nnitrogen-linked heterocyclic compounds, see: Anderson *et al.* (2004); Ha *et al.* (2009); Liu *et al.* (2011); Tan *et al.* (2012); Venkatesan *et al.* (2010); Yim *et al.* (2010). For the bioactivity of benzofuran analogues substituted by heterocyclic moieties, see: El-Shehry *et al.* (2010); Kaldrikyan *et al.* (2009); Saberi *et al.* (2006). For a related structure, see: Liu *et al.* (2012).

Experimental

Crystal data

C31H22N4O4	
$M_r = 514.53$	
Triclinic, P1	
a = 10.11 (2)	Å
b = 10.87(3)	Å
c = 11.64(3)	Å

 $\alpha = 94.73 (4)^{\circ}$ $\beta = 92.07 (3)^{\circ}$ $\gamma = 92.05 (4)^{\circ}$ $V = 1273 (5) \text{ Å}^{3}$ Z = 2Mo K\alpha radiation



 $0.38 \times 0.22 \times 0.13 \text{ mm}$

 $\mu = 0.09 \text{ mm}^{-1}$ T = 296 K

Data collection

Bruker SMART CCD area-detector	9629 measured reflections
diffractometer	4699 independent reflections
Absorption correction: multi-scan	2743 reflections with $I > 2\sigma(I)$
(SADABS; Sheldrick, 1996)	$R_{\rm int} = 0.044$
$T_{\min} = 0.966, \ T_{\max} = 0.988$	

Refinement $R[F^2 > 2\sigma(F^2)] = 0.053$

 $R[F^2 > 2\sigma(F^2)] = 0.053$ 353 parameters $wR(F^2) = 0.200$ H-atom parameters constrainedS = 0.85 $\Delta \rho_{max} = 0.26$ e Å $^{-3}$ 4699 reflections $\Delta \rho_{min} = -0.17$ e Å $^{-3}$

Table 1 Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$\begin{array}{c} C11 - H11 \cdots O3^{i} \\ C16 - H16 \cdots O3^{i} \end{array}$	0.93	2.33	3.225 (10)	161
	0.93	2.55	3.473 (10)	171

Symmetry code: (i) x, y, z - 1.

Data collection: *SMART* (Bruker, 2004); cell refinement: *SAINT* (Bruker, 2004); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

The authors are grateful for financial support by the Science and Technology Department of Zhejiang Province Foundation of China (project No. 2010 C32022) and the Zhejiang Province Natural Science Foundation of China (project No. Y4090410.)

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: DS2193).

References

- Anderson, S., Taylor, P. N. & Verschoor, G. L. B. (2004). Chem. Eur. J. 10, 518– 527.
- Bruker (2004). SMART and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
- El-Shehry, M. F., Swellema, R. H., Abu-Bakr, Sh. M. & El-Telbani, E. M. (2010). Eur. J. Med. Chem. 45, 4783–4787.
- Ha, D. T., Kim, H. J., Thuong, P. T., Ngoc, T. M., Lee, I., Nguyen, D. H. & Bae, K. H. (2009). J. Ethnopharmacol. 125, 304–309.
- Kaldrikyan, M. A., Grigoryan, L. A., Melik-Ogandzhanyan, R. G. & Arsenyan, F. G. (2009). Pharm. Chem. J. 43, 242–244.
- Liu, X. H., Tan, C. X. & Jian, Q. W. (2011). Phosphorus Sulfur Silicon Relat. Elem. 186, 552–557.

Liu, X.-H., Tan, C.-X., Weng, J.-Q. & Liu, H.-J. (2012). Acta Cryst. E68, 0493. Saberi, M. R., Vinh, T. K., Yee, S. W., Griffiths, B. J. N., Evans, P. J. & Simons, C.

- (2006). J. Med. Chem. 49, 1016–1022.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.
- Tan, C. X., Shi, Y. X., Jian, Q. W., Liu, X. H., Zhao, W. G. & Li, B. J. (2012). Lett. Drug. Des. Discov. 9, 431–435.
- Venkatesan, A. M., Santos, O. D., Ellingboe, J., Evrard, D. A., Harrison, B., Smith, D. L., Scerni, R., Hornby, G. A., Schechter, L. E. & Andree, T. H. (2010). *Bioorg. Med. Chem. Lett.* **20**, 824–827.
- Yim, N. H., Ha, D. T., Trung, T. N., Kim, J. P., Lee, S. M., Na, M. K., Jung, H. J., Kim, H. S., Kim, Y. H. & Bae, K. H. (2010). *Bioorg. Med. Chem. Lett.* 20, 1165–1168.

supplementary materials

Acta Cryst. (2012). E68, o1916 [doi:10.1107/S1600536812023720]

(5-Benzoyl-2-methyl-4-{[1-(pyridin-4-yl)-1*H*-1,2,3-triazol-4-yl]methoxy}-1benzofuran-7-yl)(phenyl)methanone

Xiao-qin Zhang, Hai-Liang Zhang, Zhu-Yong Dong, Qiang Qian and Yu-Guang Wang

Experimental

General procedure to synthesize the title compound: Under a positive pressure of nitrogen, to a suspension of the swollen 2-polystyrene supported selanylmethyl-4-(prop-2-ynyloxy)-5,7-dibenzoyl- 2,3-dihydro-benzofuran in DMSO (30 mL) was added the mixed solution of 0.2 g (4.0 mmol) Cu₂SO₄ 5H₂O and 0.8 g(4.5 mmol) ascorbic acid in 10 mL water, 5.0 mmol 4-Azido-pyridine. After stirring for 15 h at 60° C, the resin was collected by filtration, washed with H₂O (30 mL×2), THF (20 mL×1), hot DMF (15 mL×1), H2O (30 mL×1), THF (20 mL×1), THF/H₂O (2:1) (20 mL×2), hot DMF (15 mL×1), THF (20 mL×1), THF/H₂O (2:1) (20 mL×2), THF (20 mL×2). The washed resin was suspended in THF (15 mL), 30% H₂O₂ (20.0 equiv) was added, and the mixture was stirred for 10 h at room temperature. The resin was collected by filtration, washed with H_2O (20 mL×2), THF (10 mL×2), THF/ H_2O (2:1) (10 mL×2), THF (10 mL×2), CH_2Cl_2 (10 mL×2), toluene (10 mL×2). The washed resin was suspended in 15 mL toluene, DBU (0.4 equiv) was added, and the mixture was stirred for 5.0 h at 80°C. The mixture was filtered, and the resin was washed with CH₂Cl₂ (15 mL \times 2). The filtrate was washed with 0.25M HCl (30 mL \times 2), saturated sodium bicarbonate solution(35 mL \times 2), dried with anhydrous magnesium sulfate, and evaporated to dryness under vacuum to obtain the title compound. Further purification was via flash chromatography with n-hexanes-EtOAc (3:1, V/V) as the eluent for microanalyses. ¹H-NMR(CDCl₃, 400MHz, Bruker Avance spectrometer): δ 8.78-8.76 (d, 2H, J=5.6Hz), 7.85-7.81 (m, 4H), 7.64-7.40 (m, 10H), 6.77 (d, 1H, J=0.8Hz), 5.52 (s, 2H), 2.47 (s, 3H); 13C-NMR(CDCl3): δ 195.6, 192,8, 157.7, 156.2, 152.5, 151.9, 145.4, 143.1, 138.4, 137.9, 133.4, 133.2, 130.3, 130.2, 128.7, 128.6, 128.1, 125.0, 122.1, 120.5, 118.3, 113.9, 101.2, 67.1, 14.3; MS(ESI) m/z 515 (M+H)⁺. The title compound was recrystallized from CHCl₂ at room temperature to give the desired crystals suitable for single-crystal X-ray diffraction.

Refinement

All H atoms were positioned geometrically and constrained to ride on their parent atoms (N—H = 0.86 Å and $U_{iso}(H) = 1.2U_{eq}(N)$; C—H = 0.93 and 0.97 Å for aromatic and methylene H atoms with $U_{iso}(H) = 1.2U_{eq}(C)$, respectively.

Computing details

Data collection: *SMART* (Bruker, 2004); cell refinement: *SAINT* (Bruker, 2004); data reduction: *SAINT* (Bruker, 2004); program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL* (Sheldrick, 2008).





View of the title compound showing the atom numbering scheme and the ellipsoids at the 50% probability level.

(5-Benzoyl-2-methyl-4-{[1-(pyridin-4-yl)-1H-1,2,3-triazol-4-yl]methoxy}-1-benzofuran-7-yl)(phenyl)methanone

Crystal data

$C_{31}H_{22}N_4O_4$	Z = 2
$M_r = 514.53$	F(000) = 536
Triclinic, $P\overline{1}$	$D_{\rm x} = 1.343 {\rm ~Mg} {\rm ~m}^{-3}$
Hall symbol: -P 1	Mo <i>K</i> α radiation, $\lambda = 0.71073$ Å
a = 10.11 (2) Å	Cell parameters from 1346 reflections
b = 10.87 (3) Å	$\theta = 2.5 - 21.0^{\circ}$
c = 11.64 (3) Å	$\mu = 0.09 \mathrm{~mm^{-1}}$
$\alpha = 94.73 (4)^{\circ}$	T = 296 K
$\beta = 92.07 \ (3)^{\circ}$	Block, colourless
$y = 92.05 (4)^{\circ}$	$0.38 \times 0.22 \times 0.13 \text{ mm}$
V = 1273 (5) Å ³	
Data collection	

Bruker SMART CCD area-detector	9629 measured reflections
diffractometer	4699 independent reflections
Radiation source: fine-focus sealed tube	2743 reflections with $I > 2\sigma(I)$
Graphite monochromator	$R_{\rm int} = 0.044$
phi and ω scans	$\theta_{\rm max} = 25.5^{\circ}, \ \theta_{\rm min} = 2.5^{\circ}$
Absorption correction: multi-scan	$h = -12 \rightarrow 12$
(SADABS; Sheldrick, 1996)	$k = -12 \rightarrow 13$
$T_{\min} = 0.966, \ T_{\max} = 0.988$	$l = -14 \rightarrow 14$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.053$	Hydrogen site location: inferred from
$wR(F^2) = 0.200$	neighbouring sites
S = 0.85	H-atom parameters constrained
4699 reflections	$w = 1/[\sigma^2(F_o^2) + (0.1P)^2 + 0.4387P]$
353 parameters	where $P = (F_o^2 + 2F_c^2)/3$
0 restraints	$(\Delta/\sigma)_{\rm max} < 0.001$
Primary atom site location: structure-invariant	$\Delta \rho_{\rm max} = 0.26 \text{ e } \text{\AA}^{-3}$
direct methods	$\Delta \rho_{\rm min} = -0.17 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2$ sigma(F^2) is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters $(Å^2)$

	x	У	Ζ	$U_{ m iso}*/U_{ m eq}$
N1	0.4387 (2)	-0.1220 (2)	-0.3567 (2)	0.0473 (6)
N2	0.5280 (3)	-0.1333 (3)	-0.2672 (2)	0.0622 (8)
N3	0.4837 (3)	-0.0696 (3)	-0.1759 (2)	0.0614 (8)
N4	0.5183 (3)	-0.2859 (3)	-0.6907 (3)	0.0806 (10)
01	0.20278 (18)	-0.01035 (18)	-0.05837 (16)	0.0467 (5)
O2	-0.0057 (2)	-0.2440 (2)	0.12502 (19)	0.0610 (6)
O3	0.1408 (2)	0.0551 (2)	0.48995 (19)	0.0654 (7)
O4	0.19574 (18)	0.30077 (17)	0.25525 (16)	0.0464 (5)
C1	0.1656 (3)	0.1786 (3)	0.2199 (2)	0.0406 (7)
C2	0.1172 (3)	0.0928 (3)	0.2923 (2)	0.0425 (7)
C3	0.0985 (3)	-0.0274 (3)	0.2403 (2)	0.0448 (7)
H3	0.0657	-0.0882	0.2846	0.054*
C4	0.1268 (2)	-0.0616 (3)	0.1240 (2)	0.0405 (7)
C5	0.1774 (3)	0.0282 (3)	0.0540 (2)	0.0398 (7)
C6	0.1954 (2)	0.1526 (3)	0.1030 (2)	0.0401 (7)
C7	0.2451 (3)	0.2700 (3)	0.0659 (2)	0.0448 (7)
H7	0.2711	0.2844	-0.0078	0.054*
C8	0.2462 (3)	0.3540 (3)	0.1584 (2)	0.0452 (7)
C9	0.2956 (3)	0.0639 (3)	-0.1195 (3)	0.0516 (8)
H9A	0.3597	0.1067	-0.0648	0.062*
H9B	0.2482	0.1253	-0.1586	0.062*
C10	0.3658 (3)	-0.0175 (3)	-0.2056 (2)	0.0465 (7)
C11	0.3382 (3)	-0.0494 (3)	-0.3205 (3)	0.0474 (7)
H11	0.2660	-0.0265	-0.3646	0.057*
C12	0.4637 (3)	-0.1773 (3)	-0.4698 (3)	0.0491 (7)
C13	0.5661 (4)	-0.2603 (4)	-0.4842 (3)	0.0738 (11)

H13	0.6175	-0.2813	-0.4214	0.089*
C14	0.5883 (4)	-0.3104 (4)	-0.5955 (4)	0.0874 (13)
H14	0.6567	-0.3648	-0.6046	0.105*
C15	0.4212 (4)	-0.2071 (4)	-0.6724 (3)	0.0698 (10)
H15	0.3701	-0.1886	-0.7363	0.084*
C16	0.3902 (3)	-0.1506 (3)	-0.5659 (3)	0.0567 (8)
H16	0.3216	-0.0960	-0.5596	0.068*
C17	0.0912 (3)	-0.1942 (3)	0.0833 (3)	0.0468 (7)
C18	0.1739 (3)	-0.2720 (3)	0.0022 (2)	0.0448 (7)
C19	0.1132 (3)	-0.3795 (3)	-0.0558 (3)	0.0547 (8)
H19	0.0227	-0.3952	-0.0504	0.066*
C20	0.1884 (4)	-0.4627 (3)	-0.1214 (3)	0.0639 (9)
H20	0.1477	-0.5333	-0.1595	0.077*
C21	0.3240 (4)	-0.4402 (3)	-0.1299 (3)	0.0683 (10)
H21	0.3739	-0.4959	-0.1733	0.082*
C22	0.3846 (3)	-0.3340 (3)	-0.0732 (3)	0.0627 (9)
H22	0.4752	-0.3188	-0.0788	0.075*
C23	0.3098 (3)	-0.2500 (3)	-0.0078 (3)	0.0520 (8)
H23	0.3510	-0.1789	0.0293	0.062*
C24	0.0941 (3)	0.1209 (3)	0.4186 (2)	0.0456 (7)
C25	0.0118 (3)	0.2283 (3)	0.4561 (2)	0.0447 (7)
C26	-0.0814 (3)	0.2755 (3)	0.3805 (3)	0.0522 (8)
H26	-0.0883	0.2445	0.3035	0.063*
C27	-0.1641 (4)	0.3690 (3)	0.4208 (3)	0.0676 (10)
H27	-0.2274	0.3984	0.3710	0.081*
C28	-0.1517 (4)	0.4178 (3)	0.5349 (3)	0.0737 (11)
H28	-0.2062	0.4803	0.5614	0.088*
C29	-0.0573 (4)	0.3730 (3)	0.6102 (3)	0.0698 (10)
H29	-0.0478	0.4067	0.6863	0.084*
C30	0.0217 (3)	0.2784 (3)	0.5712 (3)	0.0571 (8)
H30	0.0825	0.2474	0.6221	0.068*
C31	0.2916 (4)	0.4852 (3)	0.1780 (3)	0.0660 (10)
H31A	0.3324	0.5105	0.1101	0.099*
H31B	0.3547	0.4950	0.2421	0.099*
H31C	0.2171	0.5351	0.1946	0.099*

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}	
N1	0.0469 (13)	0.0519 (15)	0.0426 (14)	0.0057 (11)	0.0023 (11)	-0.0015 (12)	
N2	0.0580 (16)	0.087 (2)	0.0416 (15)	0.0196 (15)	-0.0011 (12)	-0.0016 (14)	
N3	0.0566 (16)	0.083 (2)	0.0443 (16)	0.0135 (14)	0.0016 (12)	-0.0016 (14)	
N4	0.080 (2)	0.090 (2)	0.066 (2)	0.0052 (18)	0.0064 (17)	-0.0285 (18)	
01	0.0542 (12)	0.0487 (12)	0.0367 (11)	-0.0012 (9)	0.0065 (9)	0.0000 (9)	
02	0.0587 (13)	0.0587 (14)	0.0641 (15)	-0.0120 (11)	0.0105 (11)	-0.0002 (11)	
03	0.0742 (15)	0.0765 (16)	0.0481 (14)	0.0220 (12)	0.0009 (11)	0.0133 (12)	
04	0.0534 (12)	0.0416 (12)	0.0440 (12)	0.0014 (9)	0.0062 (9)	0.0006 (9)	
C1	0.0412 (15)	0.0398 (16)	0.0409 (16)	0.0028 (12)	0.0037 (12)	0.0013 (13)	
C2	0.0448 (15)	0.0470 (17)	0.0362 (16)	0.0024 (13)	0.0034 (12)	0.0048 (13)	
C3	0.0465 (16)	0.0443 (17)	0.0449 (17)	0.0009 (13)	0.0060 (13)	0.0093 (13)	

C4	0.0378 (14)	0.0431 (16)	0.0410 (16)	0.0015 (12)	0.0025 (12)	0.0055 (13)
C5	0.0368 (14)	0.0465 (17)	0.0357 (16)	0.0038 (12)	0.0005 (11)	0.0016 (12)
C6	0.0366 (14)	0.0433 (17)	0.0407 (16)	0.0020 (12)	0.0002 (12)	0.0054 (13)
C7	0.0474 (16)	0.0479 (18)	0.0393 (16)	0.0001 (13)	0.0063 (12)	0.0037 (13)
C8	0.0488 (16)	0.0446 (17)	0.0432 (17)	0.0023 (13)	0.0078 (13)	0.0068 (14)
C9	0.0605 (19)	0.0501 (18)	0.0438 (18)	-0.0023 (15)	0.0129 (14)	-0.0003 (14)
C10	0.0493 (17)	0.0476 (17)	0.0427 (17)	0.0017 (13)	0.0057 (13)	0.0022 (14)
C11	0.0465 (16)	0.0529 (18)	0.0430 (17)	0.0071 (14)	0.0022 (13)	0.0035 (14)
C12	0.0529 (17)	0.0502 (18)	0.0427 (17)	0.0005 (14)	0.0035 (13)	-0.0057 (14)
C13	0.072 (2)	0.080 (3)	0.066 (2)	0.026 (2)	-0.0107 (18)	-0.019 (2)
C14	0.081 (3)	0.094 (3)	0.081 (3)	0.027 (2)	-0.007 (2)	-0.038 (2)
C15	0.080(2)	0.081 (3)	0.045 (2)	0.003 (2)	0.0019 (17)	-0.0092 (18)
C16	0.062 (2)	0.061 (2)	0.0470 (19)	0.0081 (16)	0.0018 (15)	-0.0014 (15)
C17	0.0452 (16)	0.0491 (18)	0.0457 (17)	-0.0031 (14)	-0.0022 (13)	0.0055 (14)
C18	0.0489 (16)	0.0412 (17)	0.0443 (17)	0.0021 (13)	-0.0028 (13)	0.0046 (13)
C19	0.0511 (18)	0.0520 (19)	0.059 (2)	-0.0061 (15)	-0.0065 (15)	-0.0002 (16)
C20	0.080(2)	0.051 (2)	0.057 (2)	-0.0020 (18)	-0.0036 (18)	-0.0091 (16)
C21	0.080 (2)	0.067 (2)	0.057 (2)	0.0141 (19)	0.0093 (18)	-0.0092 (18)
C22	0.0550 (19)	0.061 (2)	0.073 (2)	0.0063 (16)	0.0108 (16)	0.0037 (18)
C23	0.0523 (17)	0.0480 (18)	0.0551 (19)	0.0005 (14)	0.0007 (14)	0.0027 (15)
C24	0.0467 (16)	0.0507 (18)	0.0398 (17)	-0.0002 (13)	0.0014 (13)	0.0068 (14)
C25	0.0482 (16)	0.0480 (17)	0.0388 (16)	-0.0016 (13)	0.0094 (13)	0.0067 (13)
C26	0.0605 (19)	0.0560 (19)	0.0406 (17)	0.0045 (15)	0.0101 (14)	0.0016 (14)
C27	0.074 (2)	0.070 (2)	0.062 (2)	0.0198 (19)	0.0130 (17)	0.0144 (19)
C28	0.092 (3)	0.062 (2)	0.068 (3)	0.010 (2)	0.030 (2)	-0.0012 (19)
C29	0.082 (3)	0.076 (3)	0.050 (2)	0.005 (2)	0.0135 (19)	-0.0058 (19)
C30	0.064 (2)	0.066 (2)	0.0414 (18)	0.0011 (17)	0.0065 (15)	0.0019 (15)
C31	0.085 (2)	0.050 (2)	0.062 (2)	-0.0033 (17)	0.0115 (18)	-0.0044 (17)

Geometric parameters (Å, °)

N1-C11	1.366 (4)	C13—C14	1.391 (6)
N1—N2	1.369 (4)	C13—H13	0.9300
N1-C12	1.437 (5)	C14—H14	0.9300
N2—N3	1.320 (4)	C15—C16	1.388 (5)
N3—C10	1.383 (4)	C15—H15	0.9300
N4—C15	1.337 (5)	C16—H16	0.9300
N4	1.343 (6)	C17—C18	1.509 (5)
01—C5	1.376 (4)	C18—C23	1.397 (5)
O1—C9	1.458 (4)	C18—C19	1.410 (5)
O2—C17	1.236 (4)	C19—C20	1.398 (5)
O3—C24	1.232 (4)	C19—H19	0.9300
O4—C1	1.378 (4)	C20—C21	1.392 (6)
O4—C8	1.412 (4)	C20—H20	0.9300
C1—C2	1.395 (4)	C21—C22	1.393 (5)
C1—C6	1.412 (5)	C21—H21	0.9300
С2—С3	1.398 (5)	C22—C23	1.401 (5)
C2—C24	1.505 (5)	C22—H22	0.9300
C3—C4	1.416 (5)	C23—H23	0.9300
С3—Н3	0.9300	C24—C25	1.503 (5)

C4—C5	1.416 (4)	C25—C30	1.402 (5)
C4—C17	1.507 (5)	C25—C26	1.405 (5)
C5—C6	1.426 (5)	C26—C27	1.401 (5)
C6—C7	1.459 (5)	С26—Н26	0.9300
C7—C8	1.352 (4)	C27—C28	1.389 (6)
С7—Н7	0.9300	С27—Н27	0.9300
C8—C31	1.480 (5)	C28—C29	1.399 (6)
C9—C10	1.497 (4)	C28—H28	0.9300
С9—Н9А	0.9700	C29—C30	1.382 (5)
С9—Н9В	0.9700	С29—Н29	0.9300
C10—C11	1.369 (5)	С30—Н30	0.9300
C11—H11	0.9300	C31—H31A	0.9600
C12—C16	1.376 (5)	C31—H31B	0.9600
C12—C13	1.402 (5)	C31—H31C	0.9600
C11—N1—N2	110.2 (3)	N4—C15—H15	117.3
C11—N1—C12	130.0 (3)	С16—С15—Н15	117.3
N2—N1—C12	119.7 (3)	C12-C16-C15	118.3 (3)
N3—N2—N1	106.9 (3)	C12—C16—H16	120.8
$N_2 - N_3 - C_{10}$	109.4(3)	C15—C16—H16	120.8
C15 - N4 - C14	1150(3)	02-C17-C4	117.9(3)
$C_{5} - C_{9}$	118 3 (2)	02 - C17 - C18	117.5(3)
$C_1 - C_4 - C_8$	106.1(2)	C4-C17-C18	1235(3)
04-C1-C2	100.1(2) 123.9(3)	C_{23} C_{18} C_{19}	123.3(3)
04-C1-C6	125.7(3) 110.6(2)	C_{23} C_{18} C_{17}	110.0(3) 123 1(3)
C_{2} C_{1} C_{6}	110.0(2) 125.5(3)	$C_{23} = C_{13} = C_{17}$	125.1(3) 117.8(3)
C_{1} C_{2} C_{3}	125.5(3) 114.7(3)	$C_{10} = C_{10} = C_{17}$	117.0(3)
$C_1 = C_2 = C_3$	114.7(3) 124.6(3)	$C_{20} = C_{19} = C_{18}$	120.4 (3)
$C_1 = C_2 = C_2 + C_3 + C_2 + C_3 + C_2 + C_3 $	124.0(3) 120.7(3)	$C_{20} = C_{19} = 1119$	119.8
$C_{3} = C_{2} = C_{4}$	120.7(3) 123.5(2)	$C_{10} - C_{10} - C_{10}$	119.0 120.2(2)
$C_2 = C_3 = C_4$	125.5 (5)	$C_{21} = C_{20} = C_{19}$	120.5 (5)
$C_2 = C_3 = H_3$	110.2	$C_{21} = C_{20} = H_{20}$	119.0
$C_4 = C_5 = C_4 = C_2$	110.2	C19 - C20 - H20	119.0
$C_{5} = C_{4} = C_{5}$	119.9 (3)	$C_{22} = C_{21} = C_{20}$	119.7 (5)
C_{3} C_{4} C_{17}	125.1 (3)	C22—C21—H21	120.2
C_{3} C_{4} C_{1}	114.9 (2)	$C_{20} = C_{21} = H_{21}$	120.2
01 - 05 - 04	117.3 (3)	$C_{21} = C_{22} = C_{23}$	120.3 (3)
01 - 05 - 06	124.3 (3)	C21—C22—H22	119.8
C4 - C5 - C6	118.3 (3)	C23—C22—H22	119.8
C1 - C6 - C5	118.1 (3)	C18 - C23 - C22	120.5 (3)
C1C6C7	104.9 (3)	C18—C23—H23	119.7
C5—C6—C7	136.9 (3)	С22—С23—Н23	119.7
C8—C7—C6	107.4 (3)	03-C24-C25	120.7 (3)
С8—С7—Н7	126.3	03—C24—C2	120.1 (3)
С6—С7—Н7	126.3	C25—C24—C2	119.3 (3)
C7—C8—O4	110.9 (3)	C30—C25—C26	118.7 (3)
C7—C8—C31	133.4 (3)	C30—C25—C24	119.5 (3)
O4—C8—C31	115.6 (3)	C26—C25—C24	121.7 (3)
O1—C9—C10	109.8 (3)	C27—C26—C25	120.2 (3)
O1—C9—H9A	109.7	C27—C26—H26	119.9

С10—С9—Н9А	109.7	C25—C26—H26	119.9
O1—C9—H9B	109.7	C28—C27—C26	120.1 (3)
С10—С9—Н9В	109.7	C28—C27—H27	120.0
H9A—C9—H9B	108.2	С26—С27—Н27	120.0
C11—C10—N3	107.9 (3)	C27—C28—C29	120.1 (4)
C11—C10—C9	131.1 (3)	C27—C28—H28	120.0
N3—C10—C9	121.0 (3)	C29—C28—H28	120.0
N1-C11-C10	105.6 (3)	C30—C29—C28	119.8 (3)
N1—C11—H11	127.2	С30—С29—Н29	120.1
C10-C11-H11	127.2	С28—С29—Н29	120.1
C16—C12—C13	118.5 (3)	C29—C30—C25	121.2 (3)
C16—C12—N1	121.7 (3)	С29—С30—Н30	119.4
C13—C12—N1	119.8 (3)	С25—С30—Н30	119.4
C14—C13—C12	117.9 (3)	C8—C31—H31A	109.5
C14—C13—H13	121.0	C8—C31—H31B	109.5
C12—C13—H13	121.0	H31A—C31—H31B	109.5
N4—C14—C13	124.9 (4)	C8—C31—H31C	109.5
N4—C14—H14	117.6	H31A—C31—H31C	109.5
C13—C14—H14	117.6	H31B—C31—H31C	109.5
N4—C15—C16	125.4 (4)		
C11—N1—N2—N3	0.5 (3)	N2—N1—C12—C16	-168.3 (3)
C12—N1—N2—N3	177.1 (3)	C11—N1—C12—C13	-172.8 (3)
N1—N2—N3—C10	0.3 (4)	N2—N1—C12—C13	11.3 (4)
C8-04-C1-C2	177.6 (2)	C16—C12—C13—C14	0.5 (5)
C8—O4—C1—C6	-0.1 (3)	N1—C12—C13—C14	-179.0 (3)
O4—C1—C2—C3	-177.4 (2)	C15—N4—C14—C13	0.0 (6)
C6-C1-C2-C3	0.0 (4)	C12-C13-C14-N4	-0.5 (7)
O4—C1—C2—C24	-1.4 (4)	C14—N4—C15—C16	0.6 (6)
C6-C1-C2-C24	176.0 (3)	C13—C12—C16—C15	-0.1 (5)
C1—C2—C3—C4	0.5 (4)	N1-C12-C16-C15	179.5 (3)
C24—C2—C3—C4	-175.7 (2)	N4-C15-C16-C12	-0.5 (6)
C2—C3—C4—C5	0.3 (4)	C5—C4—C17—O2	-144.0 (3)
C2—C3—C4—C17	-176.2 (2)	C3—C4—C17—O2	32.3 (4)
C9—O1—C5—C4	-158.5 (2)	C5—C4—C17—C18	41.3 (4)
C9—O1—C5—C6	23.3 (4)	C3—C4—C17—C18	-142.4 (3)
C3—C4—C5—O1	-179.9 (2)	O2—C17—C18—C23	-149.1 (3)
C17—C4—C5—O1	-3.7 (4)	C4—C17—C18—C23	25.6 (4)
C3—C4—C5—C6	-1.6 (4)	O2—C17—C18—C19	23.6 (4)
C17—C4—C5—C6	174.6 (2)	C4—C17—C18—C19	-161.7 (3)
O4—C1—C6—C5	176.4 (2)	C23—C18—C19—C20	0.4 (5)
C2-C1-C6-C5	-1.3 (4)	C17—C18—C19—C20	-172.6 (3)
O4—C1—C6—C7	-1.1 (3)	C18—C19—C20—C21	0.1 (5)
C2-C1-C6-C7	-178.7 (3)	C19—C20—C21—C22	-0.3 (5)
O1—C5—C6—C1	-179.8 (2)	C20—C21—C22—C23	0.0 (5)
C4—C5—C6—C1	2.0 (4)	C19—C18—C23—C22	-0.7 (5)
O1—C5—C6—C7	-3.4 (5)	C17—C18—C23—C22	171.9 (3)
C4—C5—C6—C7	178.4 (3)	C21—C22—C23—C18	0.5 (5)
C1—C6—C7—C8	1.9 (3)	C1—C2—C24—O3	-128.6 (3)

C5_C6_C7_C8	-1749(3)	$C_{3} - C_{2} - C_{24} - O_{3}$	471(4)
$C_{2} = C_{1} = C_{2} = C_{2}$	1/4.9(3)	C_{1} C_{2} C_{24} C_{3}	52 4 (4)
0-0/-04	-2.0(3)	C1 - C2 - C24 - C23	32.4 (4)
C6—C7—C8—C31	176.2 (3)	C3—C2—C24—C25	-131.8 (3)
C1O4C7	1.3 (3)	O3—C24—C25—C30	20.2 (4)
C1O4C8C31	-177.2 (2)	C2-C24-C25-C30	-160.8 (3)
C5-01-C9-C10	149.1 (2)	O3—C24—C25—C26	-155.7 (3)
N2-N3-C10-C11	-0.9 (4)	C2-C24-C25-C26	23.2 (4)
N2—N3—C10—C9	-177.8 (3)	C30—C25—C26—C27	-1.2 (4)
O1-C9-C10-C11	93.6 (4)	C24—C25—C26—C27	174.8 (3)
O1-C9-C10-N3	-90.4 (4)	C25—C26—C27—C28	1.8 (5)
N2-N1-C11-C10	-1.0 (3)	C26—C27—C28—C29	-0.5 (5)
C12—N1—C11—C10	-177.2 (3)	C27—C28—C29—C30	-1.2 (6)
N3-C10-C11-N1	1.2 (3)	C28—C29—C30—C25	1.8 (5)
C9—C10—C11—N1	177.6 (3)	C26—C25—C30—C29	-0.5 (5)
C11—N1—C12—C16	7.6 (5)	C24—C25—C30—C29	-176.6 (3)

Hydrogen-bond geometry (Å, °)

D—H···A	D—H	H···A	D··· A	D—H··· A
C11—H11···O3 ⁱ	0.93	2.33	3.225 (10)	161
C16—H16…O3 ⁱ	0.93	2.55	3.473 (10)	171

Symmetry code: (i) x, y, z-1.